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Use of Quasi-one-dimensional Transition Metal Ternary

Compounds and Quasi-one-dimensional Transition Metal

Chalcogenide Compounds as Electron Emitters

The object of the invention is the use of quasi-one-dimensional transition metal ternary compounds $M_xH_yHa_z$ (where M is a transition metal Mo, W, Ta, Nb; H is sulfur (S), selenium (Se), tellurium (Te); Ha is iodine (I)) and of doped quasi-one-dimensional transition metal ternary compounds $M_xH_yHa_z$ (where M=Ta, Ti, Nb; H is sulfur (S), selenium (Se), tellurium (Te); Ha is iodine (I)) with elements of group 1b (silver (Ag), gold (Au), copper (Cu)) as electron emitters under the influence of an external electric field, i.e. for cold electron emission ("cold cathode").

Electron emitters are used in various commercially available devices. Conventional emitters, used in cathode tubes, are usually made of tungsten wire or other materials with a low surface work function that emit electrons when heated (Shan I, Physics Word, 45, June (1997). Lately, vast research is being conducted on the so-called cold electron emitters which release electrons due to electric field intensity. Known devices based on cold electron emission may essentially assume two geometric forms. The first form consists in very sharp electron-emitting tips arranged in a specific configuration, this being achieved through elaborate photolithographic techniques. Said tips are generally made of silicon, molybdenum or tungsten, although recently, considerable progress has been made in the use of

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diamond powder with a specific lattice configuration, or a coating of such diamond powder or of similarly structured carbon on other microtips (Kumar N. et al., U.S. Pat. No. 5,199,918 of 04. 06. 1993). Additionally, it is possible to use microtips shaped as thin wires and scales of various materials including carbon (Q. Wang et al., App. Phys. Lett. 70, 24, pp. 3308 (1997). With the first method, highly complex lithographic techniques are required to fabricate the tips. Moreover, tips not made of diamond have a relatively short life span due to resistive heating, which gives rise to tip erosion. Both problems can be overcome by making use of diamond microtips, for proper operation, however, high anode voltage is required.

According to the second method, the device based on electron emission relies on low or negative electron affinity of the surface which is usually made of diamond or of diamond-like carbon (Kumar N. et al., U.S. Pat. No. 5,341,063 of 08. 23. 1994; Valone S. N. et al., U.S. Pat. No. 5,602,439 of 02. 11. 1997). The second method likewise requires a high anode voltage which rather complicates the operation of the device. To improve functionality, diamond diamond-like carbon with various lattice-structure defects have been employed (Jaskie J. E. et al., U.S. Pat. No. 5,619,092 of 04. 08. 1997) and some other improvements have been introduced (Habermann et al., J. Vac. Sci. Tech. B16, p. 693, 1998; Patterson D. E. et al., Mat. Res. Soc. Symp. Proc. 509, 1998).

In recent years new nanomaterials, particularly carbon and $B_xC_yN_z$ nanotubes have been found to constitute a promising source for electron emission (Zettl A. et al., U.S. Pat. No. 6,057,637 of 05. 02. 2000). Due to the shape

of these nanomaterials as such the stability of the cathode current is good as opposed to metal tips which lose sharpness over time, and work voltages are lower as compared to diamond powder. There are still several problems hampering their application, however — primarily the non-homogeneity of the material due to the uncontrolled synthesis and the problem of the disposition of the nanotubes with respect to the substrate (De Heer W. A. et al., Science 270, 1179 (1995)), although there have been attempts to reduce these disadvantages with new methods of synthesis.

Therefore, the following drawbacks are exhibited by conventional cold electron emitters: the wear and tear of the metal tip, the elaborate and costly tip fabrication technique wherein a diamond is required, the high anode voltage, and, for nanomaterials, the low definition or reproducibility of the emitters.

It is the object of the present invention to disclose and utilize such materials for fabricating cold electron emitters that will allow emitters to function in a stable and lasting manner, to be readily manufactured, and to operate at low anode voltages.

According to the invention, said object is achieved by employing quasi-one-dimensional transition metal ternary compounds and quasi-one-dimensional transition metal chalcogenide compounds as electron emitters, as defined by the independent patent claims.

The invention shall now be described with reference to two preferred embodiments thereof and illustrated in the accompanying drawings, in which:

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Fig. 1 schematically shows a measuring cell for measuring cold emission;

Fig. 2 shows the material made of bundles of $MoS_{2-y}I_x$ nanotubes which is employed as the electron emission source in an external electric field;

Fig. 3 is a typical picture on the screen created by electrons emitted from bundles of $MoS_{2-y}I_x$ nanotubes;

Fig. 4 is a diagram showing the voltage dependence of the emission current from $\text{MoS}_{2-\nu}I_{\kappa}$;

Fig. 5 shows the time stability of the emission current of bundles of $MoS_{2-y}I_x$ nanotubes;

Fig. 6 shows the material made of $Ag_x(NbS_4)_4I_y$ used as electron emission source in an external electric field;

Fig. 7 is a typical picture on the screen created by electrons emitted from $Ag_x(NbS_4)_4I_y$;

Fig. 8 is a diagram showing the voltage dependence of the emission current from $Ag_x(NbS_4)_4I_y$;

Fig. 9 shows the time stability of the emission current from $Ag_x(NbS_4)_4I_y$.

The present invention relates to the use of quasi-one-dimensional transition metal ternary compounds $M_xH_yHa_z$ (where M is a transition metal Mo, W, Ta, Nb; H is sulfur (S), selenium (Se), tellurium (Te); Ha is iodine (I)) and of doped quasi-one-dimensional transition metal ternary compounds $M_xH_yHa_z$ (where M=Ta, Ti, Nb; H is sulfur (S), selenium (Se), tellurium (Te); Ha is iodine (I)) with elements of group 1b (silver (Ag), gold (Au), or copper (Cu)) as electron emitters under the influence of an external electric field. The percentage of quasi-one-dimensional transition metal ternary compounds and/or doped quasi-one-dimensional transition metal ternary compounds

doped with elements of group 1b in the active material ranges from 0.01 to 99.9 %, the rest consisting of additives in the form of conducting, non-conducting or semi-conducting compounds or composites.

In accordance with the invention said materials are employed as the cathode emitter material in devices based on electron emission under the influence of an external electric field. Electron emission takes place at a pressure below 1 mbar.

First Embodiment

Use of bundles of $\text{MoS}_{2-y} I_{\mathrm{x}}$ nanotubes for electron emission

Bundles of $MoS_{2-y}I_x$ nanotubes are used as cold emitter. The emission characteristic of the bundles of $MoS_{2-y}I_x$ nanotubes (that is, the emission current I in dependence of the voltage U) employed as the cold electron emitter (cathode) is measured in high vacuum. The measurement of cold emission in a measuring cell of the type used in our research is schematically depicted in Figure 1. It is based on a glass vacuum vessel with two metal electrodes. The measuring cell consists of a protective resistor 1, anode 2 with a screen, an ion-getter pump 3, a nano-emitter 4, a voltage source 5, and an ammeter 6. The protection resistor is a 33 MOhm carbon (graphite-ceramic) resistor. The anode 2 is the positive electrode which at the same time serves as the screen in the FEM, thus allowing the pattern produced by the emitted electrons to be observed, said screen consisting of a 5-micrometers-thick layer of granular luminescent substance (the average grain size

being 1.2 micrometers). The luminescent substance labeled P-43 and its chemical structure is Gd_2O_2 : Tb, coated with an aluminum evaporation layer 90 nm thick. The iongetter pump 3 is a vacuum pump based on discharge in a gas and diffusion. For the measurement, a Leybold Heraeus IZ30 pump was used. The nano-emitter 4 is. the negative electrode, namely the cathode in the FEM. In embodiment the nano-emitter consists of bundles of $MoS_{2-y}I_{x}$ nanotubes. Both the nano-emitter 4 and the anode 2 are in a high vacuum. As voltage supply 5, a precision stabilized rectifier Thorn EMI, model PM 28B was used. The anode voltage was varied in the range of 600 to 2800 $\rm V.$ current was measured with a Keithley model 485 picoammeter 5. The eye symbol designates the direction of observation of the emitted light.

The bundles of $MoS_{2-y}I_x$ nanotubes were attached on the 0.1 mm thick Ni sheet with silver paste which ensures good electrical contact with the conducting substrate. The brand used was SPI Flash dry, 4999AB paste, supplied by SPI Supplies for the preparation of samples for scanning electron microscopy. Α sample of bundles of $MoS_{2-v}I_x$ nanotubes with a mass of approx. 3 mg is shown in Figure 2. It was applied on the tip of the metal conductor by means of the same paste, at a distance a from the anode. Prior to measuring the emission, the sample was pumped for several hours at room temperature, and subsequently for another 3 hours at 200° C. The current-voltage characteristic and the emission stability of the thus prepared sample measured at room temperature. The first tests performed at a distance $a=20\pm1$ mm, wherein it was possible to visually observe the emission pattern at high currents.

With this measurement it was confirmed that the measured current is definitely a consequence of cold emission through vacuum and not a consequence of a leaking insulation. The typical picture on the screen created by electrons emitted from the bundles of $MoS_{2-y}I_x$ nanotubes is shown in Figure 3. The voltage was 5 kV. The subsequent measurements of the voltage dependence of the current and of the time dependence of the current were carried out at a distance $a=5\pm1$ mm. The anode voltage was varied in the range of 600 to 2800 V. The current was measured with a picoammeter 6.

The readout was recorded with a personal computer 2.8 times per second. At the maximum voltage of the power supply 3000 V the voltage on the emitter was 2770 V, whereas the emission current reached the value of 1 μA as shown in Figure 4. The measurement of the time stability of the emission current over a period of 16 hours showed the emission of electrons from the bundles of $MoS_{2-y}I_x$ nanotubes to be relatively stable, Figure 5.

Second Embodiment

Use of $Ag_x(NbS_4)_4I_y$ for electron emission

As cold emitter $Ag_x(NbS_4)_4I_y$ is used. The emission characteristic of $Ag_x(NbS_4)_4I_y$ (that is, the emission current I in dependence of the voltage U) employed as the cold electron emitter is measured in high vacuum. The measurement of cold emission as used in our research is schematically shown in Figure 1 and is analogous to the first embodiment.

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4 mg of $Ag_x(NbS_4)_4I_v$ as shown in Figure 6 were first attached to a 0.1 mm Ni foil by means of a conducting binder (SPI, silver paste for SEM), whereupon the foil was glued on the tip of the metal conductor using the same adhesive, at a distance a from the anode 2. Prior to measuring the emission, the sample was pumped for several hours in the FEM at room temperature, and subsequently for 3 hours another at 200° C. The current-voltage characteristic and the emission stability of the thus prepared sample were measured at room temperature. first tests were performed at a distance $a=20\pm1$ mm, where it was possible to visually observe the emission pattern at high currents. With this measurement it was confirmed that the measured current is definitely a consequence of cold emission through vacuum and not a consequence of a leaking insulation. The typical picture on the screen created by electrons emitted from the bundles of Agx(NbS4)4Iv nanotubes is shown in Figure 7.

The subsequent measurements of the voltage dependence of the current and of the time dependence of the current were carried out at a distance $a=5\pm1$ mm. The anode voltage was varied in the range of 600 to 2800 V, as can be seen in Figure 8. The current was measured with a picoammeter 6. The readout was recorded with a personal computer 2.8 times per second. At 2800 V the voltage on the emitter was 1750 V, whereas the emission current reached the value of 30 μ A. The measurement of the time stability of the emission current over a period of 330 hours showed the emission of electrons from the Ag_x(NbS₄)₄I_y material to be relatively stable.

Materials based on quasi-one-dimensional transition metal ternary compounds $M_xH_yHa_z$ (where M is a transition metal Mo, W, Ta, Nb; H is sulfur (S), selenium (Se), tellurium (Te); Ha is iodine (I)) and/or doped quasi-onedimensional transition metal ternary compounds $M_xH_vHa_z$ (where M=Ta, Ti, Nb; H is sulfur (S), selenium (Se), tellurium (Te); Ha is iodine (I)) doped with elements of group 1b (silver (Ag), gold (Au), or copper (Cu)) are thus used as electron emitters under the influence of external electric field. The percentage of quasi-onedimensional transition metal ternary compounds and/or doped quasi-one-dimensional transition metal ternary compounds doped with elements of group 1b in the active material ranges from 0.01 to 99.9 %, the rest consisting additives in the form of conducting, non-conducting or semi-conducting compounds or composites. Electron emission takes place at a pressure below 1 mbar.

It is to be understood that electron emitters under the influence of an external electric field that are made of materials based on quasi-one-dimensional transition metal ternary compounds $M_x H_y H a_z$ (where M is a transition metal Mo, W, Ta, Nb; H is sulfur (S), selenium (Se), tellurium (Te); Ha is iodine (I)) and/or doped quasi-onedimensional transition metal ternary compounds $M_xH_yHa_z$ (where M=Ta, Ti, Nb; H is sulfur (S), selenium (Se), tellurium (Te); Ha is iodine (I)) doped with elements of group 1b (silver (Ag), gold (Au), or copper (Cu)), wherein the percentage of quasi-one-dimensional transition metal ternary compounds and/or doped quasi-one-dimensional transition metal ternary compounds doped with elements of group 1b in the active material ranges from 0.01 to 99.9 %,

the rest consisting of additives in the form of conducting, non-conducting or semi-conducting compounds or composites are likewise considered to be within the scope of the present invention.